

POWERING SPACE EXPLORATION: Np-237 TARGET MANUFACTURING



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Method

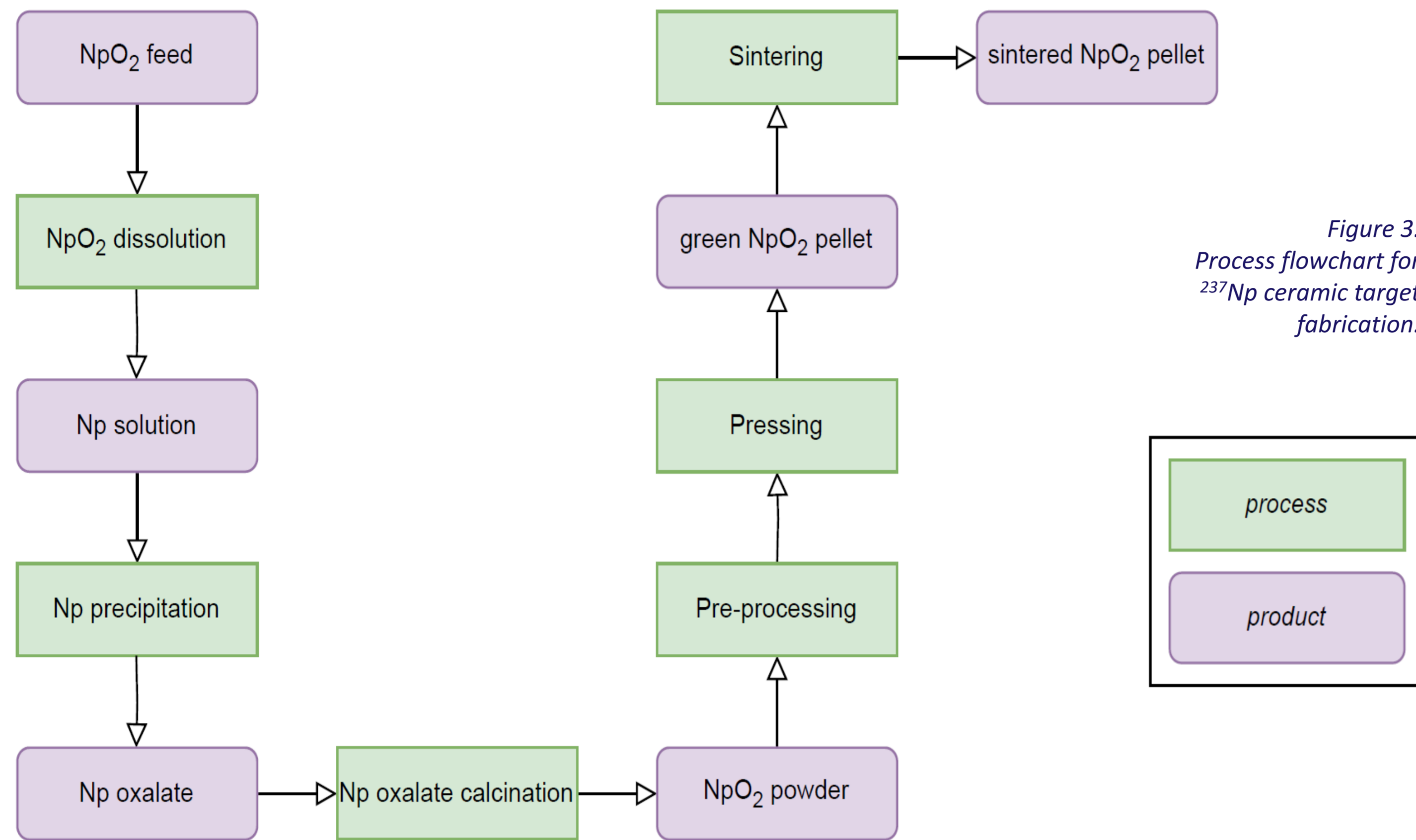


Figure 3.
Process flowchart for
²³⁷Np ceramic target
fabrication.

Introduction

Europe strives for ²³⁸Pu

For exploration of the outer parts of our planetary system, solar panels combined with batteries are no longer a standalone energy source option. NASA's deep-space exploration probes and planetary rovers rely on radioisotope heating units (RHU) and thermoelectric generators (RTG) powered by Plutonium-238 (²³⁸Pu). ²³⁸Pu is the preferred isotope since it is an alpha emitter with an almost total absence of penetrating gamma radiation. It has a convenient half-life of 86.4 years combining high specific power density with steady power supply for decades. ²³⁸Pu is produced through high-flux neutron irradiation of ²³⁷Np.

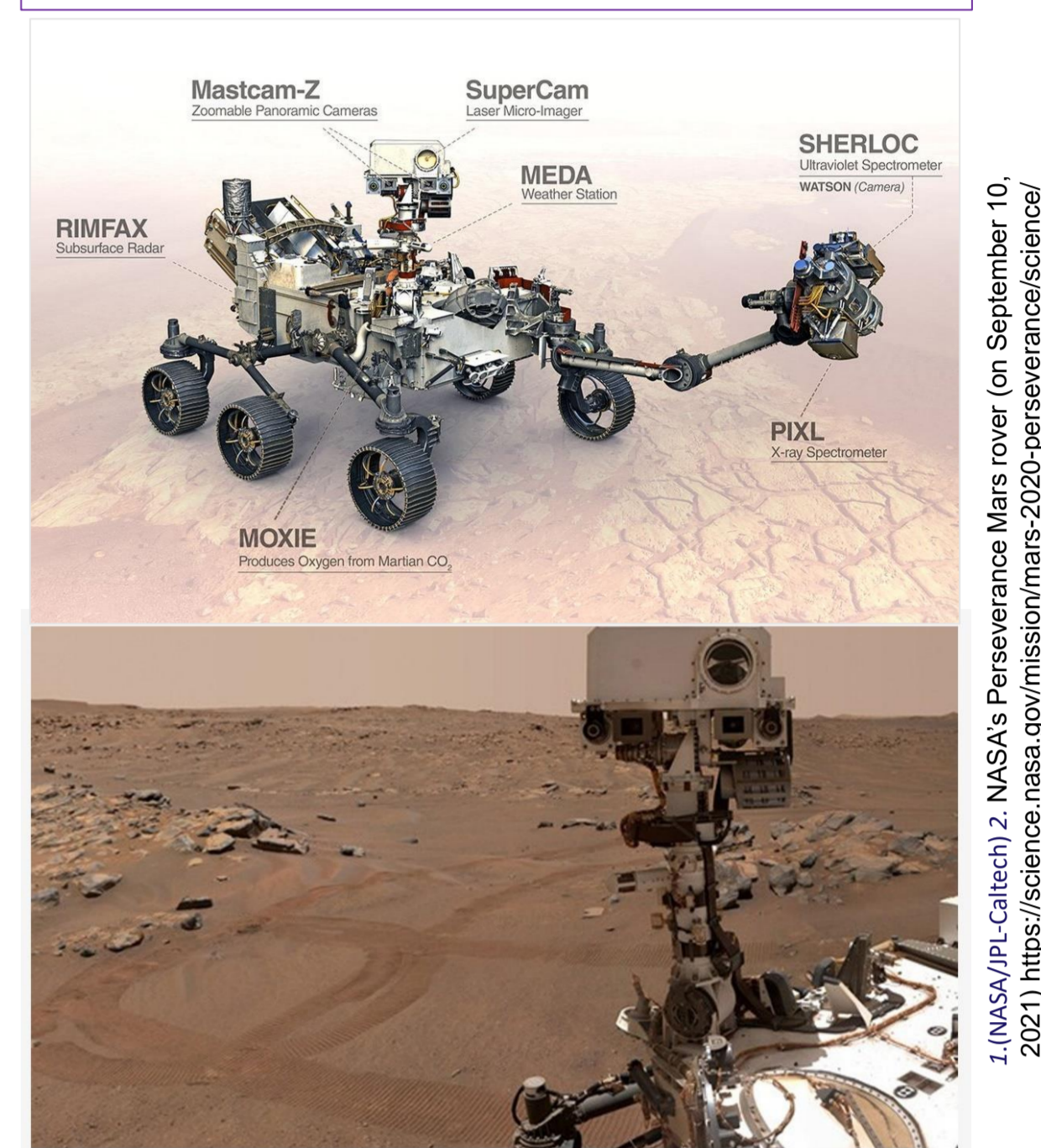
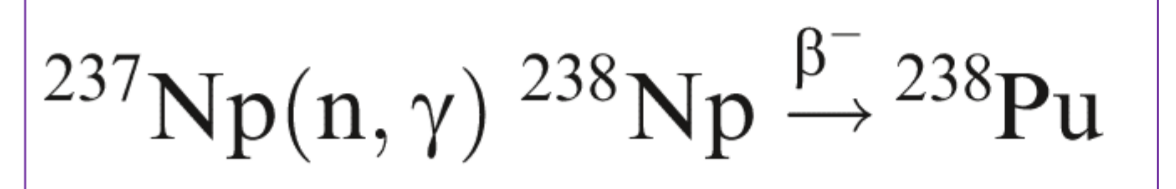


Figure 1. Perseverance Mars rover's scheme and "selfie".

Given the advantages of ²³⁸Pu, there is a renewed interest to re-open a European ²³⁸Pu production chain for its deep space exploration missions.

Objectives

Optimus Pro project contributes to outlining a way of producing ²³⁸Pu in Europe. It will demonstrate fabrication of ²³⁷Np targets at laboratory scale.

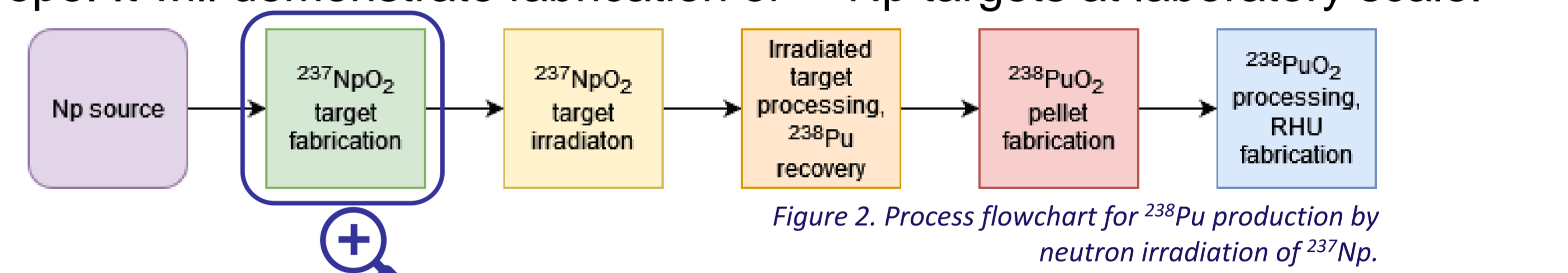
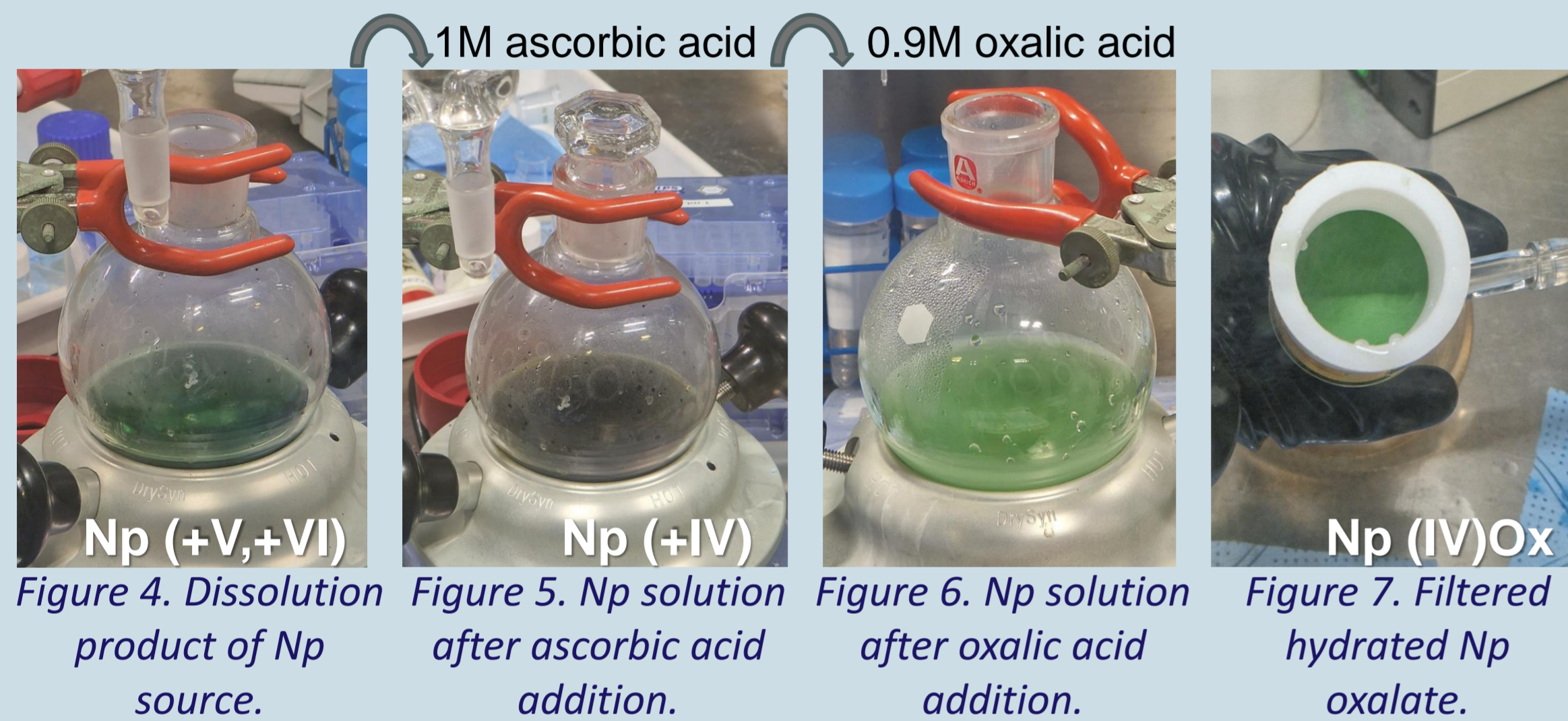
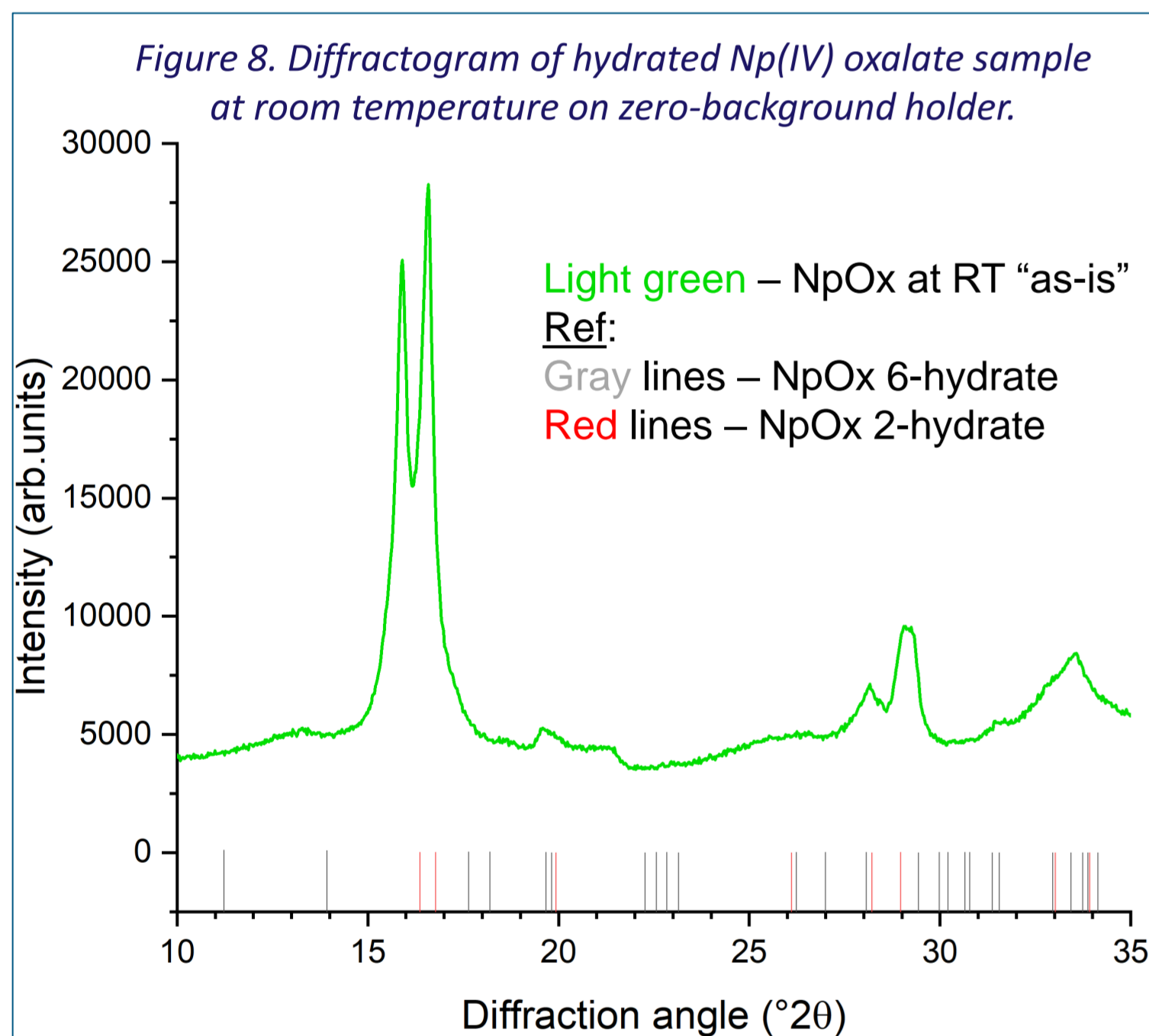


Figure 2. Process flowchart for ²³⁸Pu production by neutron irradiation of ²³⁷Np.



Np can be converted to solid phase via oxalic precipitation. It requires a full conversion of Np to the +IV oxidation state. Ascorbic acid is used for the reduction from Np(+V, +VI) to +IV. Np(IV) oxalate forms a non-soluble precipitate in aqueous solutions.



Oxalates upon heating in air undergo degradation to oxide. Powder X-ray diffraction and thermogravimetry were used to determine the conversion conditions.

²³⁷Np Oxalate production and characterisation

²³⁷Np Oxide characterisation and processing

Characterisation of obtained NpO₂ powder contributes to pressing and sintering conditions elucidation.

Measured weight fraction	Calculated weight fraction of NpO ₂ to Np(C ₂ O ₄) ₂ · nH ₂ O, where						
	n = 6	n = 5	n = 4	n = 3	n = 2	n = 1	n = 0
51.5 %	51.6 %	53.5 %	55.5 %	57.6 %	59.9 %	62.4 %	65.1 %

Table 1 - Weight fraction (%) of the TG product (NpO₂) with respect to the initial compound; measured and calculated values assuming various oxalate hydrates.

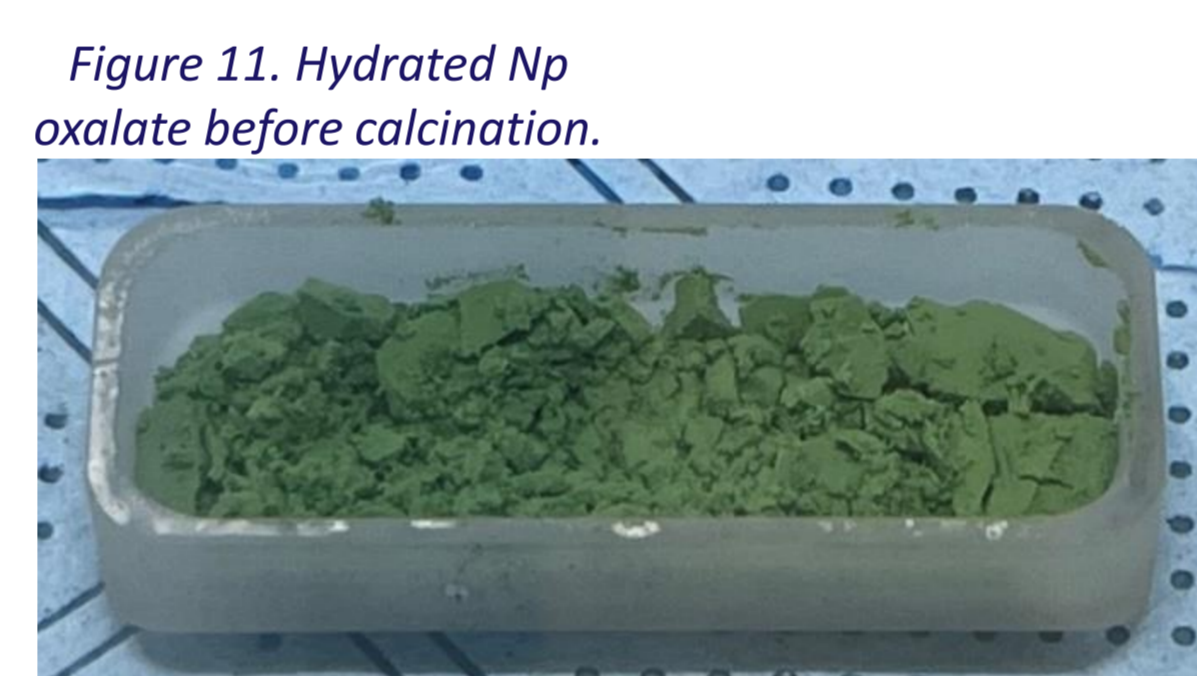


Figure 10.
Thermogravimetric curve of hydrated Np(IV) oxalate sample.

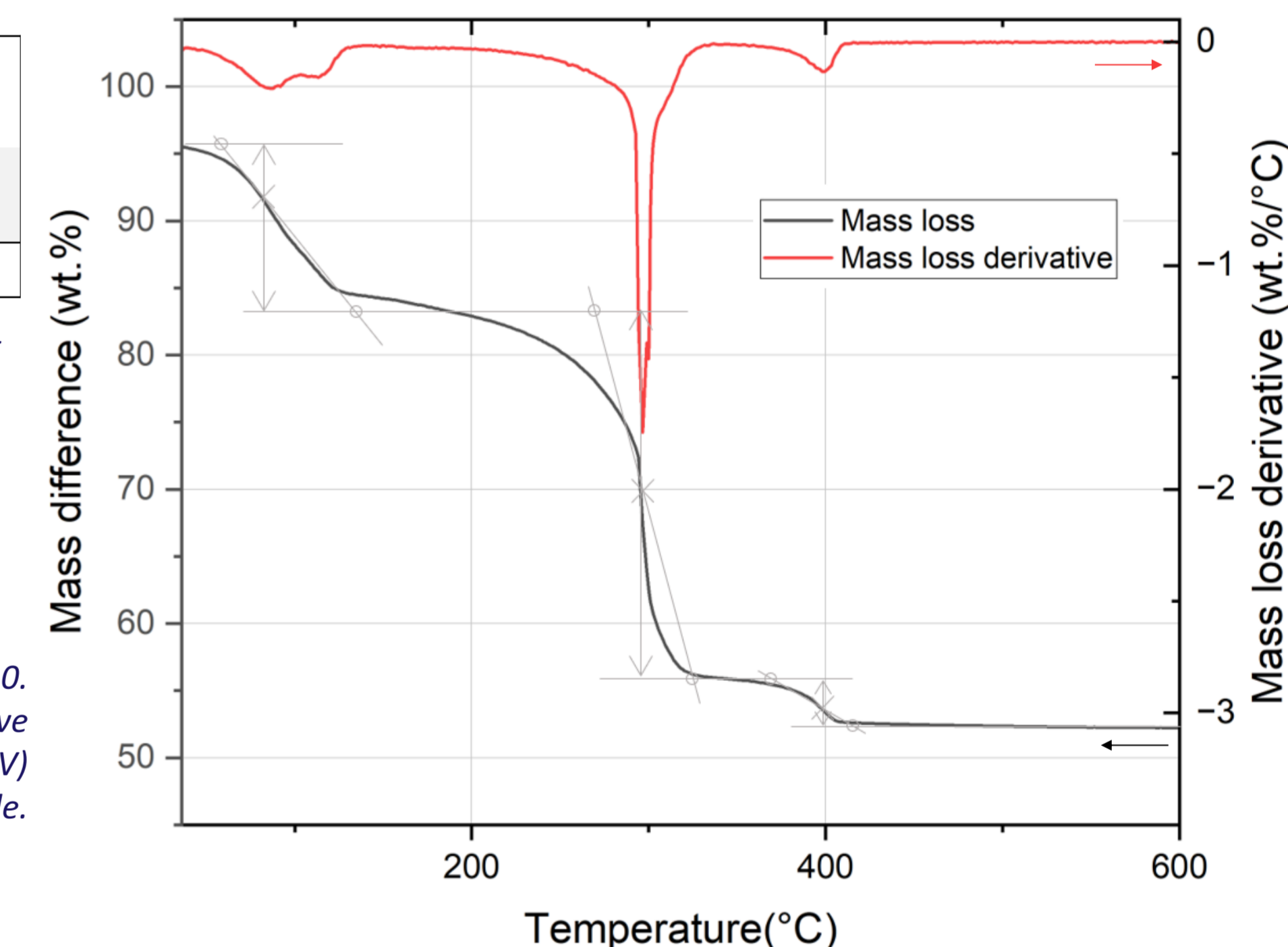


Figure 12. Hydrated Np oxalate after calcination.

550 °C

Figure 13.
Scanning electron micrographs of produced Np oxide, at various magnifications.

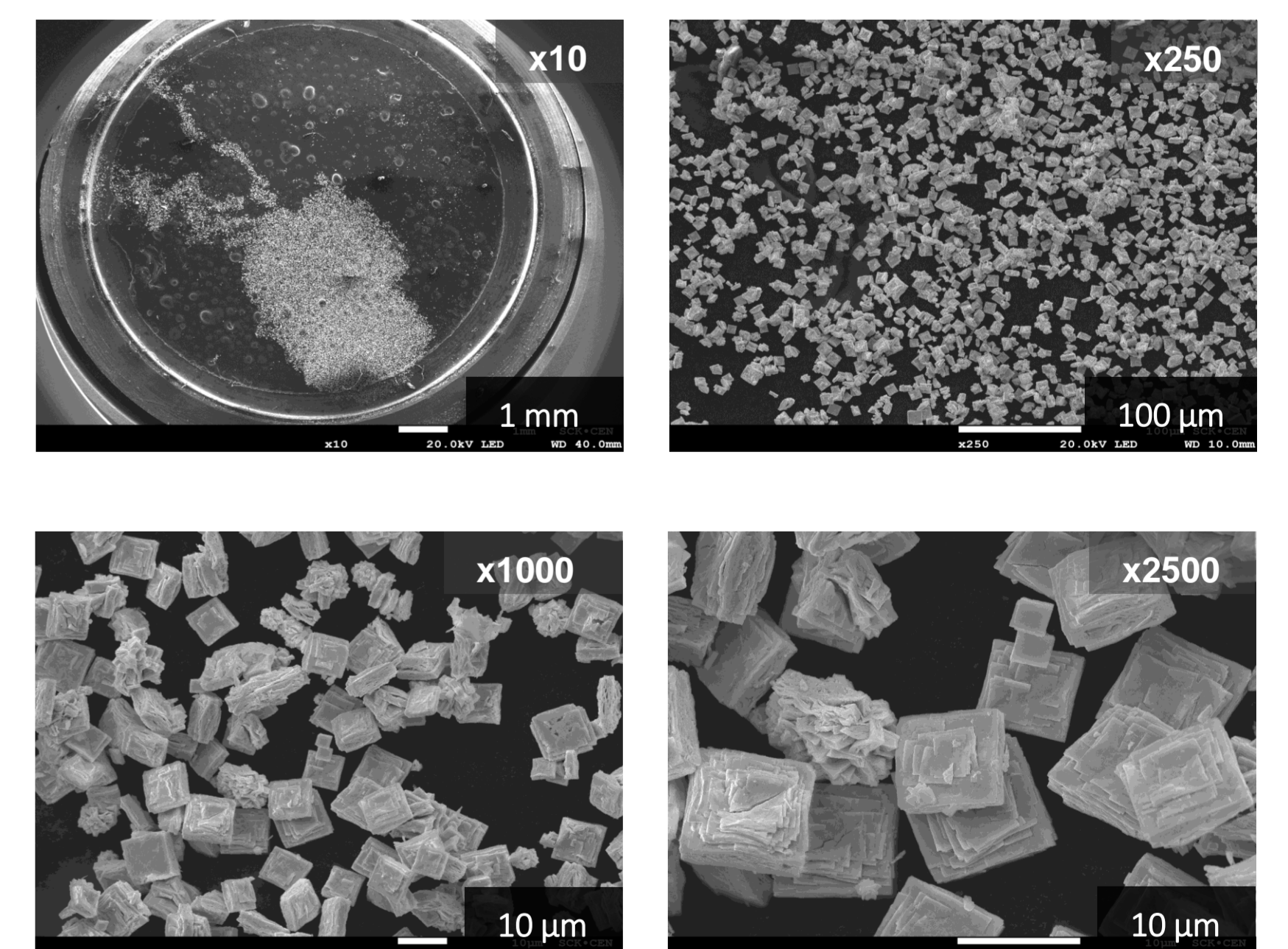


Table 2. Brunauer-Emmett-Teller method assessment of specific surface area of Np oxide powder.

BET surface area (m ² · g ⁻¹)	C	Correlation coefficient	Sampling mass (g)	Total surface
P/P ₀ = 0.05 - 0.3				
3.446 ± 0.012	102	0.999950	1.1318	3.9



Figure 14. Photos of NpO₂ pressed pellet in different light conditions.

Conclusion

The demonstration of fabrication process yielding dense NpO₂ pellets in Europe within Optimus Pro project represents a significant advancement in the development of an independent ²³⁸Pu production chain, supporting future deep-space missions.

Acknowledgements

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